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*equation do I*

*use? 11. Kinetic*

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ANSWER Answer

the following

questions in the

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b Pressure —  
orce. For a  
constant force,  
when the surface  
area is tripled  
the surface area  
pressure is (a)  
doubled. as  
much. (c ripled.  
7-0 (d)  
unchanged. Rank  
the following  
pressures in  
increasing

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order. (c) 76  
torr (a) 50 kPa  
0, 00ictbv-x  
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If a gas and a liquid are the same temperature and pressure, diffusion occurs much faster in the gas because.

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### Review Gases

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A. there are more elastic collisions

between the

particles in a

gas. B. gases

are more

compressible. C.

the particles

move faster in a

gas and there is

a greater

distance between

them.



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Dalton's Law of  
Partial

Pressures Goals

To describe the  
properties of  
mixtures of

gases. To

describe

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Calculations

that deal with  
mixtures of  
gases. In the

real world,

gases are

usually

mixtures. This

section

describes how

mixing gases

affects the

properties of

the resulting

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mixture. Gases

Section 1

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Gases

Discovering the  
Relationships

Between

Properties If we  
want to explain  
why a weather  
balloon carrying  
instruments into

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the upper Gases

atmosphere

expands as it

rises, we need

to consider

changes in the

properties of

the gases

(pressure,

volume,

temperature, or

number of gas

particles)

inside and

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outside the

balloon.

Section 1

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ANSWER Answer

the following

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### 1. b Pressure

surface area. For a constant force,

when the surface area is tripled

the pressure is

(a) doubled. (b)

a third as much.

(c) tripled. (d)

unchanged. 2. d,

c, a, b Rank the

following

pressures in

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increasing

order. (a) 50  
kPa (c) 76 torr  
(b) 2 atm (d)  
100 N/m<sup>2</sup> 3.

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this theory  
explains some of  
the properties  
of ideal gases.



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In this chapter, you will study the predictions of kinetic-molecular theory for gases in more detail. This includes the relationship among the temperature, pressure, volume, and amount of gas in

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a sample. Review Gases

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Terms

pressure newton

barometer

millimeters of

mercury

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the following  
questions in the  
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1. State whether  
the pressure of  
a fixed mass of  
gas will  
increase,  
decrease, or  
stay the same in  
the following

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circumstances:

increase a.

temperature

increases,

volume stays the

same decrease b.

volume

increases,

temperature

stays the same

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the following

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Pressure = \_\_\_\_\_. For  
a constant  
force, when the  
surface area is  
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doubled. (b) a  
third as much.  
(c) tripled. (d)  
unchanged. 2.

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