

## Chemistry Unit 7 Reaction Equations Ws1 Answers

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Unit 7: Chemical Equations Page 7 Types of Chemical Reactions Notes Synthesis- two or more \_\_\_\_ or \_\_\_\_ combine to form one compound. o Usually releases energy, \_\_\_\_ o 4 Fe (s) + 3 O 2 (g) ? 2 Fe 2 O 3 (s) CaO (s) + H 2O (l) ? Ca(OH) 2 (s) Decomposition- a single compound \_\_\_\_ into two or more elements or

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Modeling Chemistry 1 U7 WS1 2013 Name Date Pd Chemistry - Unit 7- Chemical Reactions Balancing Reaction Equations Worksheet 1 Balance the following equations by inserting the proper coefficients. You must keep track of the total number of elements on each side of the equation by keeping a tally beneath each reaction.

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2H 2(g) + O 2(g) ? 2H 2O(g) In the symbolic equation, chemical formulas are used instead of chemical names for reactants and products and symbols are used to indicate the phase of each substance. It should be apparent that the chemical shorthand method is the quickest and clearest method for writing chemical equations.

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is a chemical reaction where one reactant is exchanged for one ion of a second reactant. Single displacement reactions take the form. A + BC ? B + AC. Double Replacement Reaction. is a chemical reaction where two reactant ionic compounds exchange ions to form two new product compounds with the same ions.

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Unit 7 Chemical Equations and Reactions What is a Chemical Equation? A Chemical Equation is a written representation of the process that occurs in a chemical reaction. A chemical equation is written with the Reactants on the left side of an arrow and the Products of the chemical reaction on the right side of the equation.

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