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Bond. Molecule.

Diatomic molecule.

Molecular

compound. A bond

formed by the

sharing of electrons

between atoms. A

neutral group of

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atoms joined
together by
covalent bonds. A
molecule consisting
of two atoms.

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of water shows

how the oxygen ...

How are atoms

joined together to

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Vocabulary.

Covalent bond.

Molecule. Diatomic molecule.

Molecular compound. bond

formed by the sharing of electrons between atoms. a neutral group of atoms joined together by covalent bonds. a

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molecule that
contains two
atoms. a
compound
composed of
molecules.

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Bonding □ In

covalent bonds,

atoms share

electrons. □ There

are several

electrostatic

interactions in

these bonds:

Attractions

between electrons

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between electrons,

Repulsions

between nuclei.

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general, covalent
bonding results

from an imbalance
between the

attractions and
repulsions of the
nuclei and

electrons involved.

□ The nuclei and

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electrons attract
each other. □

Nuclei repel other
nuclei.

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Section Review

Objectives □ State
a rule that usually
tells how many

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Education are

shared to form a
covalent bond □

Describe how
electron dot

formulas are used

□ Predict when two
atoms are likely to

be joined by a
double or a triple
covalent bond □

Distinguish

between a single
covalent bond and

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other covalent
bonds

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Covalent bonding is
a form of chemical
bonding between
two non metallic
atoms which is
characterized by

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the sharing of pairs
of electrons
between

Covalent

Bonding

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Rule4. □ Atoms

tend to gain, lose,

or share electrons

until they are

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surrounded by eight valence electrons; this is known as the octet rule. □ An octet consists of full sand p subshells. □ We know that ns^2np^6 is a noble gas configuration.

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