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Solution From Coulomb's law of electrostatic attraction we know that the PE between two charges Q1 and Q2 separated by a distance r is given by QQ PE = 4πϵ o r First we consider the interaction between Na+ ion and 6Cl- ions at distance r Applying Coulomb's law we have Q1Q2 (-6e)(+e) = - 6e PE1 = = = 4πϵ o r 4πϵ o r 4πϵ o r Similarly, we now consider 12 Na+ ions as second nearest neighbors at a distance r QQ (+12e)(+ e) 12e = PE2 = = 4πϵ o r 4πϵ o r 4πϵ o r and Cl ...

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Ernest Rutherford (1871-1937) 1.37 fSolutions to Principles of Electronic Materials and Devices: 2nd Edition (Summer 2001) Chapter 2 Second Edition ( 2001 McGraw-Hill) Chapter 2 2.1 Electrical conduction Na is a monovalent metal (BCC) with a density of 0.9712 g cm-3. Its atomic mass is 22.99 g mol-1.

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