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Particles, Gamma

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Neutrons!

**Calculating the
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Neutrons and
Electrons for an
Atom Calculating**

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Neutrons**

The Periodic
Table Song |

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Basic Parts of
the Atom -

Protons,

Neutrons,

Electrons,

Nucleus **How to**

**Find the Charge
of an Ion! (The
Octet Rule)**

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What are

Protons,

Neutrons and

Electrons?

~~Atomic Number,~~

~~Atomic Mass, and~~

~~the Atomic~~

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in the blanks in
the following
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Please keep in
mind that the
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represented by
each space may
NOT be the most
common isotope
or the one
closest in
atomic mass to

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Neutrons And
the value on the
periodic table.

Atomic symbol

Atomic number.

Answers

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symbol Atomic

number Protons

Neutrons

Electrons Atomic

mass Charge Pb

82 82 125 80 207

+2 Se 34 34 45

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34 79 0 Cr 24 24

28 21 52 +3 F 9

9 10 9 19 0 Nb

41 41 52 35 93

+5 P 15 15 16 18

31 -3 Rb 37 37

48 36 85 +1

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Electrons Atomic
Charge Protons
mass

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= Atomic Number.

Electrons =

Protons. #

Neutrons =

Atomic Mass -

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Atomic Number OR

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table to find

the numbers of

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Mass Number

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understand how

to project cash

flow. Whatever

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planning goals,
cash flow is the
resource in the
company, and
managing cash is
the one small
business
purpose. Version
control is
another issue
with Excel.

Protons Neutrons

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Electrons Atomic

and Mass

Worksheet

Answers

The number of

neutrons in an

atom is not

fixed same

elements may

have atoms with

different number

of neutrons.

Oxygen has the

atomic number 8

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and an atomic mass of 16 that means it has 8 protons 8

electrons and 8 neutrons rules for calculating protons

electrons and neutrons in uncharged compounds protons.

Therefore an element in a

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Are protons and
neutrons the
same number
Electrons are
...

Atoms are the
basic building
blocks of
ordinary matter.
They have the

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ability to join
together and
form molecules,
which also join
together to form
the object that
we can clearly
see around us.

Atoms are
composed of
particles called
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neutrons. Take

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